







Collective Effects in Equilibrium and Nonequilibrium Physics: June 19, 2006	5
Turing on Broken Symmetry	
There appears superficially to be a difficulty confronting this theory of morphogenesis, or, indeed, almost any other theory of it. An embryo in its spherical blastula stage has spherical symmetry But a system which has spherical symmetry, and whose state is changing because of chemical reactions and diffusion, will remain spherically symmetrical for ever It certainly cannot result in an organism such as a horse, which is not spherically symmetrical.	
There is a fallacy in this argument. It was assumed that the deviations from spherical symmetry in the blastula could be ignored because it makes no particular difference what form of asymmetry there is. It is, however, important that there are <i>some</i> deviations, for the system may reach a state of instability in which these irregularities, or certain components of them, tend to growIn practice, however, the presence of irregularities, including statistical fluctuations in the numbers of molecules undergoing the various reaction, will, if the system has an appropriate kind of instability, result in this homogeneity disappearing.	

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Collective Effects in Equilibrium and Nonequilibrium Physics: June 19, 2006 6 Modern Interpretation • We now know that the structural information of biological organisms is encoded at the molecular level in the DNA • Coding is in terms of base sequences that code for the production of proteins with a rate controlled by other proteins: these can be thought of as the morphogens • How does the information at the molecular level become structure at the macroscopic level? • This process obviously uses the laws of physics, but are the "laws" of pattern formation involved? ♦ Yes in modelling \diamond ?? in real world Back Forward

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Reaction-Diffusion

Two chemical species with concentrations u_1, u_2 that react and diffuse

$$\partial_t u_1 = f_1 (u_1, u_2) + D_1 \partial_x^2 u_1$$

 $\partial_t u_2 = f_2 (u_1, u_2) + D_2 \partial_x^2 u_2$

• Reaction:

$$a \mathbf{A} + b \mathbf{B} \rightarrow c \mathbf{C} + d \mathbf{D}$$

gives the reaction rate (law of mass action)

$$v(t) = -\frac{1}{a} \frac{d[\mathbf{A}]}{dt} = \dots = k[\mathbf{A}]^{m_{\mathbf{A}}} [\mathbf{B}]^{m_{\mathbf{B}}}$$

with $m_A = a \dots$ for elementary reactions

• Diffusion: conservation equation

 $\partial_t u_i = -\nabla \cdot \mathbf{j}_i$

with

$$\mathbf{j}_i = -D_i \nabla u_i$$

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"Reaction" and "Diffusion"	
The form of the equations is not actually specific to chemical systems: nonlinear local terms and second order derivatives appear in many other systems:	
• nerve fibres (Hodgkin-Huxley), heart tissue etc.	
 reaction: currents across membrane through ion-channels with dynamic gate variables 	
♦ diffusion: resistive flow of current along membrane	
neural networks	
$\diamond~$ reaction: neuron firing as nonlinear function of inputs	
♦ diffusion: connectivity	
• gene networks	
\diamond reaction: gene expression controlled by other gene products	
\diamond diffusion: transport of gene products from cell to cell	
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Turing Instability

• Stationary uniform base solution $\mathbf{u}_b = (u_{1b}, u_{2b})$

$$f_1(u_{1b}, u_{2b}) = 0$$
$$f_2(u_{1b}, u_{2b}) = 0$$

• Linearize about the base state $\mathbf{u} = \mathbf{u}_b + \delta \mathbf{u}$

 $\partial_t \delta u_1 = a_{11} \delta u_1 + a_{12} \delta u_2 + D_1 \partial_x^2 \delta u_1$

$$\partial_t \delta u_2 = a_{21} \delta u_1 + a_{22} \delta u_2 + D_2 \partial_x^2 \delta u_2$$

with $a_i = \partial f_i / \partial u_j \big|_{\mathbf{u} = \mathbf{u}_b}$.

• Seek a solution $\delta \mathbf{u}(t, x)$ that is a Fourier mode with exponential time dependence:

$$\delta \mathbf{u} = \delta \mathbf{u}_q \ e^{\sigma_q t} \ e^{iqx} = \begin{pmatrix} \delta u_{1q} \\ \delta u_{2q} \end{pmatrix} e^{\sigma_q t} e^{iqx}$$

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Collective Effects in Equilibrium and Nonequilibrium Physics: June 19, 2006 Stability Analysis • Eigenvalue equation $\mathbf{A}_q \, \delta \mathbf{u}_q = \sigma_q \, \delta \mathbf{u}_q$ where $\mathbf{A}_q = \begin{pmatrix} a_{11} - D_1 q^2 & a_{12} \\ a_{21} & a_{22} - D_2 q^2 \end{pmatrix}$ • Eigenvalues are $\sigma_q = \frac{1}{2} \text{tr} \mathbf{A}_q \pm \frac{1}{2} \sqrt{(\text{tr} \mathbf{A}_q)^2 - 4 \det \mathbf{A}_q}$

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Gene Circuit Description (Reinitz et al.)

Equations for the concentrations v_i^a of gene product *a* at nucleus *i*

$$\dot{v}_{i}^{a} = R^{a} g_{a} \left(\sum_{b} T^{ab} v_{i}^{b} + m^{a} v_{i}^{bcd} + h^{a} \right) + D(n) \left[(v_{i-1}^{a} - v_{i}^{a}) + (v_{i+1}^{a} - v_{i}^{a}) \right]$$

reaction diffusion

• Reaction: nonlinear interaction term

$$g_a(u) = \frac{1}{2} \left[1 + \frac{u}{\sqrt{u^2 + 1}} \right], \qquad g_a(-\infty) = 0, \ g_a(\infty) = 1$$

with T^{ab} an interaction matrix, v_i^{bcd} the (fixed) concentration of the maternal *bcd* gene product, and R^a , m^a , h^a constants. (Other formulations replace g_a by a binary on-off function.)

- Diffusion: discrete nuclei-nuclei transport with diffusion constant D(n) depending on cleavage cycle n (geometry)
- The many parameters are fit over several cleavage cycles to large data base

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